Note-Taking Guide: Light and the Big Bang

Learning Target: I can construct an explanation of the Big Bang theory based on astronomical evidence of light spectra, motion of distant galaxies, and composition of matter in the universe.

Types of Electromagnetic Radiation		
0	Electromagnetic can be characterized by their wavelength or frequency	
	and classified into different types of radiation.	
0	The wavelengths are called radio waves, followed by microwaves,	
	infrared, visible light, ultraviolet, x rays, and gamma rays.	
0	The wavelength of light is related to its frequency by the wave equation.	

Type	Wavelength		
	> 0.	3 m	
	0.3 m	1 mm	
	1 mm	789 nm	
	789 nm	625 nm	
	625 nm	600 nm	
	600 nm	577 nm	
	577 nm	491 nm	
	455 nm	390 nm	
	390 nm	8.82 nm	
	8.82 nm	6 pm	
	< 6 pm		

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•	Frequency -	Energy	ке	iations	nın

- The energy of the radiation is related to the frequency of the light wave:
- The constant *h* is known as the Planck constant, and it is equal to:
- What is an electron volt? An **electron volt** is the energy required to accelerate an electron across an electric _______ of 1 volt.
- Think, Pair, Share: What are some common sources of light? What do they have in common?

at a given time is defined as its temperature . All matter with a temperature aboveemits thermal radia. Thermal Radiation Curve Thermal radiation consists of a wide range of wavelengths—due to a wide spectrum ofeven at a A few atoms move very fast— wavelength radiation. A few atoms move very slow— wavelength radiation. The majority of atoms move at a speed that depends on the temperature of the object. For different temperatures, an object will emit thermal radiation that different wavelengths. wavelengths. Wavelength of Maximum Intensity		in matter is known as thermal radiation .
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·		wavelengths.
o The wavelength of maximum intensity is inversely proportional to the temperature of	Wavele	ength of Maximum Intensity
	0	The wavelength of maximum intensity is inversely proportional to the temperature of the
object measured in Kelvins (K):		object measured in Kelvins (K):

Example 2: The Sun appears to give off radiation with a wavelength of maximum intensity at 508 nm. What is the temperature of the Sun's surface?

 Observa 	tion of Light with Specific Wavelengths
0	If a low-density gas is given energy, it will also produce light—but in a different way.
0	Each has specific wavelengths of light that it emits after being given
	energy.
• The Boh	r Model of an Atom
0	The Bohr model of the atom explains this phenomenon.
0	The Bohr model predicts that the in atoms are in orbits of differing
	energy (called "energy levels") around the nucleus.
0	The electron can be in energy level or another but nothing in between.
0	The electron normally occupies the lowest energy level—or the
	—but it can be "" to a higher energy level.
0	After being excited, the electron can return to its original ground state by
	the energy it absorbed.
0	The energy is released as a of electromagnetic radiation, and the
	amount of energy is equal to
Observations of t	
	Abundance of Elements
0	Astronomers look at the Sun's radiation and they find that it is made mostly of
	, some, and traces of other gases.
0	Based on the light from stars and galaxies, the universe appears to be made of approximately Hydrogen, Helium, and less than 0.1% other elements.
0	A theory of the origin of the universe must account for this abundance of hydrogen and
	helium.
• Cosmic I	Microwave Background
0	In 1964, while attempting to detect faint radio waves bounced off balloon satellites, Arno
	Penzias and Robert Wilson kept getting interference (like static) in their receiver.
0	After attempting to get rid of all possible sources of interference—including removing bird
	droppings—they determined that the radio noise was coming from, and
	in all directions.
0	The radiation they were detecting looked like a thermal radiation curve with a peak
	wavelength of 1.07 mm. What does this imply about the temperature of the universe?