

Battery and Bulb Investigations

Simple Experiments in Electricity

Name: _____

Today you will be investigating how to connect “batteries” and light bulbs to make the bulbs light in different ways. (The word “battery” technically means more than one electrochemical cell and what we typically call a “AA battery” is only a single cell, so it’s really a “AA cell” which becomes a “battery” when used in conjunction with other cells. We won’t make a big deal of this though—just wanted you to know.)

Equipment Needed:

2 D cells	2 blue light bulb sockets
1 AA cell	2 blue D cell holders
2 flashlight bulbs	4 red <i>or</i> black wires (with some alligator clips)

Please do not return equipment in a haphazard way—the same person that obtains the equipment is responsible for returning it—in the same or better condition!

Do not connect more than two cells with our flashlight bulbs—the tiny filaments inside will melt and break.

Procedure

- 1) Take *one* D-cell (not in its blue holder), *one* wire, and *one* bulb (not in the blue socket) and make the bulb light. Draw a diagram of the working configuration below.

- 2) Take one AA-cell, one wire, and one bulb (not in the blue socket) and make the bulb light. Draw a diagram of the working configuration below.

What do you notice about the brightness of the bulb now as compared in step 1? (Repeat step 1 if necessary.)

What does this suggest about the two types of cells?

